

A New Approach for Ingenol Synthesis

Tsuyoshi Nakamura, Takuya Matsui,
Keiji Tanino,* and Isao Kuwajima*

Department of Chemistry, Tokyo Institute of Technology,
Meguro, Tokyo 152, Japan

Received January 30, 1997

Ingenol¹ (**1**), isolated from the genus *Euphorbia*, has been of great interest as a synthetic target² because of its unusual structure involving an “inside-outside” bridged BC ring coupled with a broad spectrum of biological activities (Figure 1).³ Construction of the highly strained *trans*-bridged BC ring system of ingenol through a direct cyclization reaction has proved difficult, and therefore, multistep transformations have been required.^{2f,h,p} In the present paper, we describe a new entry to ingenane skeleton synthesis via a tandem cyclization–rearrangement strategy.⁴

The *trans*-decalin derivative, including the side chain with a dicobalt hexacarbonyl propargyl cation,⁵ was designed as a precursor of tandem cyclization–rearrangement reactions (Scheme 1). The intramolecular electrophilic addition of the propargyl cation moiety to the ethylidene carbon would afford the tricyclic tertiary cation intermediate that, in turn, undergoes rearrangement to yield the desired ingenane skeleton.

This strategy has several advantages: (1) The conformational rigidity of the *trans*-decalin framework as well as the dicobalt hexacarbonyl propargyl cation moiety would facilitate the cyclization of the C ring. (2) The *trans*-diaxial relationship between Ha and the hydroxy group of the decalinol would be advantageous for induc-

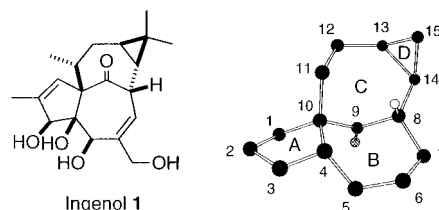
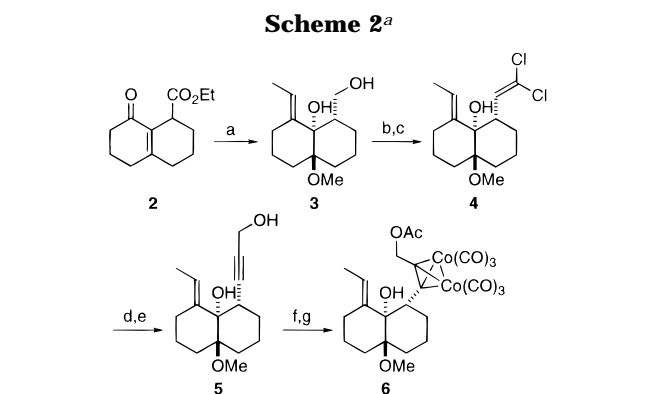
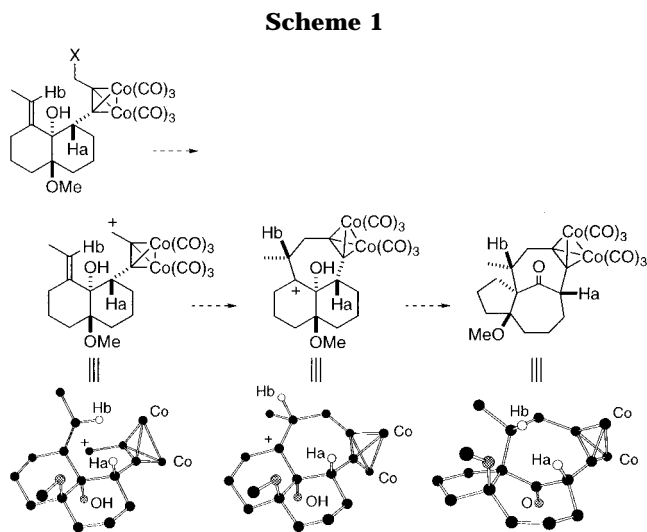


Figure 1.



^a Key: (a) Supporting Information; (b) DMSO, (COCl)₂, Et₃N, CH₂Cl₂, 100%; (c) Cl₃CPO(OEt)₂, BuLi, THF–Et₂O, –100 °C to rt, 95%; (d) BuLi, THF, –78 °C then MeOCOCl, –45 °C; (e) DIBAL, toluene, –78 °C, two steps 67%; (f) Ac₂O, Et₃N, DMAP, CH₂Cl₂, 97%; (g) Co₂(CO)₈, CH₂Cl₂, rt, 98%.

ing smooth transformation to the ingenane skeleton having the unique “inside” α -proton that is almost antiparallel to the carbonyl group. (3) Stereocontrol at the C(11) position (ingenane numbering) would be achieved by using the (*E*)-isomer of the ethylidenedecalin derivative. (4) The dicobalt hexacarbonyl acetylene moiety of the product would be useful as a flag for installation of the D ring.

The ingenane skeleton precursor was prepared as shown in Scheme 2. *trans*-Decalinol **3** with three continuous chiral centers was synthesized from keto ester **2**, which was easily prepared according to a previously described procedure,⁶ through diastereoselective epoxidation followed by regioselective hydrolysis of the epoxide.⁷ Diol **3** was converted into dichloroolefin **4** via Swern oxidation and Horner–Emmons reaction. Successive treatment of **4** with excess butyllithium and methyl chloroformate afforded an acetylenic ester that was reduced to propargyl alcohol **5**. Acetylation followed

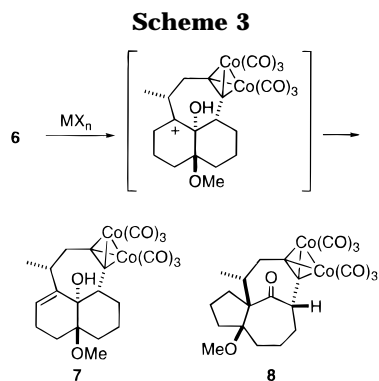
(1) (a) Zechmeister, K.; Brandl, F.; Hoppe, W.; Hecker, E.; Opferkuch, H. J.; Adolf, W. *Tetrahedron Lett.* **1970**, *47*, 4075. (b) Hecker, E. *Pure Appl. Chem.* **1977**, *49*, 1423. (c) Evans, F.; Soper, C. *Lloydia* **1978**, *41*, 193. (d) Adolf, W.; Hecker, E. *Isr. J. Chem.* **1977**, *16*, 75.

(2) (a) For a review, see: Rigby, J. H. In *Studies in Natural Products Chemistry*; Rahman, A.-u., Ed.; Elsevier: Amsterdam, 1993; Vol. 12 (part H), pp 233–74. (b) Paquette, L. A.; Nitz, T. J.; Ross, R. J.; Springer, J. P. *J. Am. Chem. Soc.* **1984**, *106*, 1446. (c) Rigby, J. H.; Moore, T. L.; Rege, S. *J. Org. Chem.* **1986**, *51*, 2398. (d) Funk, R. L.; Bolton, G. L. *J. Am. Chem. Soc.* **1986**, *108*, 4655. (e) Mehta, G.; Pathak, V. P. *J. Chem. Soc., Chem. Commun.* **1987**, 876. (f) Winkler, J. D.; Henegar, K. E. *J. Am. Chem. Soc.* **1987**, *109*, 2850. (g) Ross, R. J.; Paquette, L. A. *J. Org. Chem.* **1987**, *52*, 5497. (h) Funk, R. L.; Olmstead, J. A.; Parvez, M. *J. Am. Chem. Soc.* **1988**, *110*, 3298. (i) Rigby, J. H.; Moore, T. L. *J. Org. Chem.* **1990**, *55*, 2959. (j) Winkler, J. D.; Gretler, E. A.; Willard, P. G. *J. Org. Chem.* **1993**, *58*, 1973. (k) Funk, R. L.; Olmstead, J. A.; Parvez, M.; Stallman, B. J. *J. Org. Chem.* **1993**, *58*, 5873. (l) Rigby, J. H.; Cuisiat, S. V. *J. Org. Chem.* **1993**, *58*, 6286. (m) Winkler, J. D.; Henegar, K. E.; Hong, B.; Willard, P. G. *J. Am. Chem. Soc.* **1994**, *116*, 4183. (n) Winkler, J. D.; Hong, B.; Kim, S.; Lewin, N. E.; Blumberg, P. M. *Synlett* **1995**, 533. (o) Winkler, J. D.; Hong, B.; Buhador, A.; Kazanietz, M. G.; Blumberg, P. M. *J. Org. Chem.* **1995**, *60*, 1381. (p) Rigby, J. H.; Claire, V. S.; Cuisiat, S. V.; Heeg, M. J. *J. Org. Chem.* **1996**, *61*, 1446.

(3) (a) Hecker, E. *Cancer Res.* **1968**, *28*, 2338. (b) Kupchan, S. M.; Uchida, I.; Branfman, A. R.; Dailey, R. G.; Fei, B. Y. *Science (Washington D.C.)* **1976**, *191*, 571. (c) Jefferey, A. M.; Liskamp, R. M. *J. Proc. Natl. Acad. Sci. U.S.A.* **1986**, *83*, 241. (d) Itai, A.; Kato, Y.; Tomioka, N.; Iitaka, Y.; Endo, Y.; Hasegawa, M.; Shudo, K.; Fujiki, H.; Sakai, S. *Proc. Natl. Acad. Sci. U.S.A.* **1988**, *85*, 3688. (e) Wender, P. A.; Cribbs, C. M.; Koehler, F. K.; Sharkey, A. N.; Herald, C. L.; Kamano, Y.; Pettit, G. R.; Blumberg, P. M. *Proc. Natl. Acad. Sci. U.S.A.* **1988**, *85*, 7197. (f) Nakamura, H.; Kishi, Y.; Pajares, M. A.; Rando, R. R. *Proc. Natl. Acad. Sci. U.S.A.* **1989**, *86*, 9672.

(4) For examples of the total synthesis of natural products using a tandem cyclization–rearrangement strategy, see: Overman, L. E. *Acc. Chem. Res.* **1992**, *25*, 352.

(5) (a) Nicholas, K. M. *Acc. Chem. Res.* **1987**, *20*, 207. (b) Schreiber, S. L.; Sannakia, T.; Crowe, W. E. *J. Am. Chem. Soc.* **1986**, *108*, 3128. (c) Schreiber, S. L.; Klimas, M. T.; Sannakia, T. *J. Am. Chem. Soc.* **1987**, *109*, 5749. (d) Krafft, E. M.; Cheng, Y. Y.; Wright, C.; Cali, F. J. *J. Org. Chem.* **1996**, *61*, 3912. (e) Jamison, T. F.; Shambayati, S.; Crowe, W. E.; Shreiber, S. L. *J. Am. Chem. Soc.* **1994**, *116*, 5505.

**Table 1. Reactions of 6 with Aluminum Reagents**

entry	AlX ₃ ^a	products (%) ^b		
		7	8	6
1	Me ₂ AlCl	50		
2	Me ₂ Al(OTf)	69	13	
3	MeAl(OTf) ₂	66	13	
4	MeAl(OCOCF ₃) ₂	23	38	24
5	MeAl(OCOCF ₃)(OAr) ^c	21	77	
6	MeAl(OCOCF ₃)(OAr ²) ^c			80

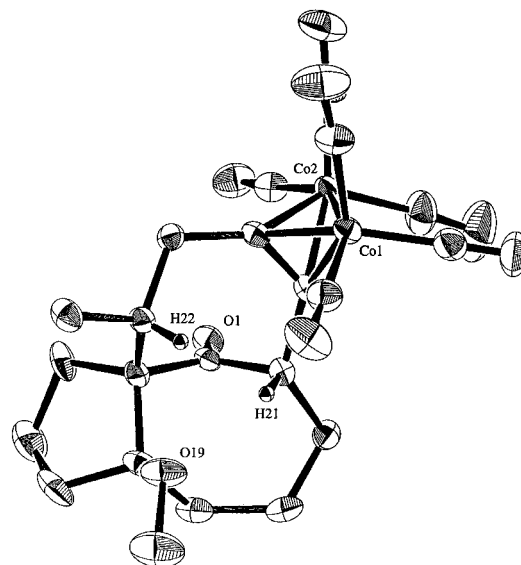
^a Aluminum reagents except for Me₂AlCl were prepared from a 1 M hexane solution of Me₃Al and the corresponding acids or phenols at room temperature. Acetate **6** was treated with 2.2 equiv of an aluminum reagent in CH₂Cl₂ at -23 °C to rt. ^b Determined by ¹H NMR of the crude product using CHBr₃ as an internal standard. ^c Ar¹ = 2,6-(CH₃)₂-4-(NO₂)C₆H₂, Ar² = 4-NO₂C₆H₄.

by complexation using Co₂(CO)₈⁵ gave dicobalt hexacarbonyl propargyl cation precursor **6**.⁸

Initial attempts to induce a tandem cyclization–rearrangement reaction of propargyl acetate **6** under the influence of an equimolar amount of Lewis acid were unsuccessful. Thus, the reactions with TiCl₄ and SnCl₄ gave complex mixtures, and only a small amount of allyl alcohol **7** was obtained in the presence of BF₃·OEt₂, EtAlCl₂, or Me₂AlCl (Scheme 3).

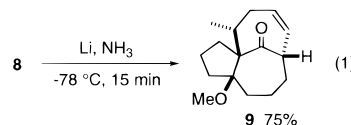
The formation of **7** indicates that the dicobalt hexacarbonyl propargyl cation attacks the ethylidene carbon to generate the tricyclic tertiary cation intermediate, which prefers β-elimination rather than rearrangement of the carbon framework. We envisioned that *in situ* activation of the hydroxy group by forming a metal alkoxide would significantly accelerate the rearrangement pathway. Use of 2 equiv of an aluminum reagent was expected to be suitable for this purpose because a variety of aluminum reagents can be easily prepared from trimethylaluminum.⁹ The reactions of **6** with several aluminum reagents are summarized in Table 1.

Although the use of Me₂AlCl was not effective (entry 1), a small amount of the desired product **8** (Figure 2)⁸ was obtained by the use of Me₂AlOTf or MeAl(OTf)₂ (entries 2 and 3), and MeAl(OCOCF₃)₂¹⁰ afforded **8** as the major product (entry 4). These results suggest that low Lewis-acidity of an aluminum reagent is essential for the pinacol-type rearrangement promoted by the aluminum alkoxide of the decalinol. After further investigation, aluminum 2,6-dimethyl-4-nitrophenoxide was found to

**Figure 2.** ORTEP drawing of cobalt complex **8**.

give the most satisfactory results (entry 5), which contrast with the reaction of aluminum 4-nitrophenoxide, which resulted in recovery of the starting material (entry 6). The methyl groups at the *o*-position of 4-nitrophenol seem to play an important role perhaps by activating the aluminum reagent by reducing aggregation.¹¹

Finally, reductive deprotection of the dicobalt hexacarbonyl acetylene moiety of **8** was examined. The hydrogenation catalyzed by Wilkinson complex¹² as well as Birch reduction using lithium metal afforded the desired product **9** having the C(13)–C(14) double bond, which would be useful for installation of the D ring via cyclopropanation reaction (eq 1).



In conclusion, we developed a new method for constructing the highly strained ingenane skeleton by a tandem cyclization–rearrangement strategy. Product **9**, which contains the C(11) α-methyl group as well as the C(13)–C(14) double bond, shows promise for the total synthesis of ingenol. We are currently investigating the synthesis and tandem cyclization–rearrangement reactions of functionalized decalinol derivatives.

Acknowledgment. The authors are grateful to Professor Hiroharu Suzuki and Dr. Masato Oshima, Department of Chemical Technology, TIT, for X-ray crystallographic structure determination. This work was partially supported by Grants from the Ministry of Education, Science, Sports, and Culture of the Japanese Government.

Supporting Information Available: Experimental procedures and characterization data for all new compounds and the ORTEP drawings of **6** and **8** (10 pages).

JO970172N

(6) Strekowski, L.; Kong, S.; Battiste, M. A. *J. Org. Chem.* **1988**, *53*, 901.

(7) *trans*-Decalinol **3** was derived in 12 steps from **2** and described fully in the Supporting Information.

(8) The stereochemistries of both **6** and **8** were eventually established by X-ray crystallographic structure determination.

(9) Zietz, J. R., Jr.; Robinson, G. C.; Lindsay, K. L. In *Comprehensive Organometallic Chemistry*; Barton, D. H. R., Ed.; Pergamon Press: Oxford, 1983; Chapter 46.

(10) Hashimoto, S.; Ito, A.; Kitazawa, H.; Yamamoto, H.; Nozaki, H. *J. Am. Chem. Soc.* **1974**, *99*, 4192.

(11) (a) Maruoka, K.; Yamamoto, H. *Angew. Chem., Int. Ed. Engl.* **1985**, *24*, 668. (b) Yamamura, Y.; Umeyama, K.; Maruoka, K.; Yamamoto, H. *Tetrahedron. Lett.* **1982**, *23*, 1933. (c) Sakane, S.; Fujiwara, J.; Maruoka, K.; Yamamoto, H. *J. Am. Chem. Soc.* **1983**, *105*, 6154.

(12) (a) Hosokawa, S.; Isobe, M. *Synlett* **1995**, 1179. (b) Isobe, M.; Yenjai, C.; Tanaka, S. *Synlett* **1994**, 916.